

<p>89-259866/36 D25 E19 G04 KAO CORP 22.01.88-JP-012061 (27.07.89) C11d-01/28 D061-01/04 Detergent compsn. for dry cleaning - contg. mixt. of sulpho- tri:carb:allylic ester salts C89-115536</p>	<p>KAOS 22.01.88 *J0 1188-595-A D(1)-A1A, 11-A1B2, 11-B16, 11-D1; E(10-A9B8) G(4-B8)</p>
<p>A new detergent compsn. for dry cleaning contains one or a mixt. of sulphotricharballylic ester salts of formulae (I) and (II)</p> $\begin{array}{c} \text{CH}_2\text{COOR}_1 \\ \\ \text{XO}_2\text{S} - \text{C} - \text{COOR}_2 \quad (\text{I}) \\ \\ \text{CH}_2\text{COOR}_2 \end{array}$ $\begin{array}{c} \text{XO}_2\text{S} - \text{CHCOOR}_1 \\ \\ \text{CHCOOR}_2 \quad (\text{II}) \\ \\ \text{CH}_2\text{COOR}_3 \end{array}$ <p>R_1, R_2 and R_3 = 1-2C opt. branched alkyl or alkenyl, or one of H, alkali metals, alkaline earth metals and org. amines (at least two of them are the alkyl or alkenyl gps.); and X = alkali metal alkali earth metal, or org amine.</p>	<p><u>USE/ADVANTAGE</u> The compsn. has good phase stability and a good water-solubilising ability (a hydrated ability). It thus readily forms a uniform pretreating liq with any commercial petroleum solvent.</p> <p><u>EXAMPLE</u> (I) and (II) include the sodium salts of tri-2-ethyl butyl sulphotricharballylate, trihexyl sulphotricharballylate and tri-2-ethyl hexyl sulphotricharballyate, and the potassium salts of trioctyl sulphotricharballylate, trioctyl sulphotricharballylate, and tributyl sulphotricharballylate. (6ppW31MBDwgNo. 0/0).</p> <p>J00188595-A</p>